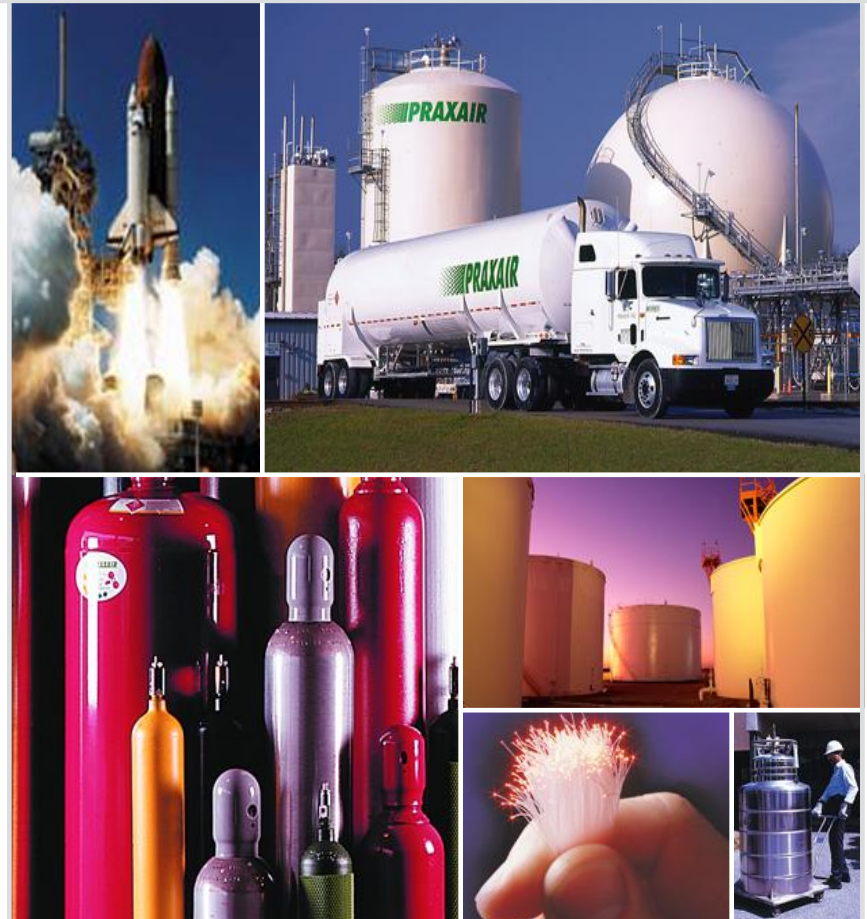




Near Zero Emissions Oxy-combustion Flue Gas Purification

Minish Shah, Nick Degenstein, Monica
Zanfir, Ravi Kumar, Jennifer
Bugayong and Ken Burgers
2010 NETL CO₂ Capture Technology
Meeting, Pittsburgh, PA
September 13 – 17, 2010



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Praxair At A Glance

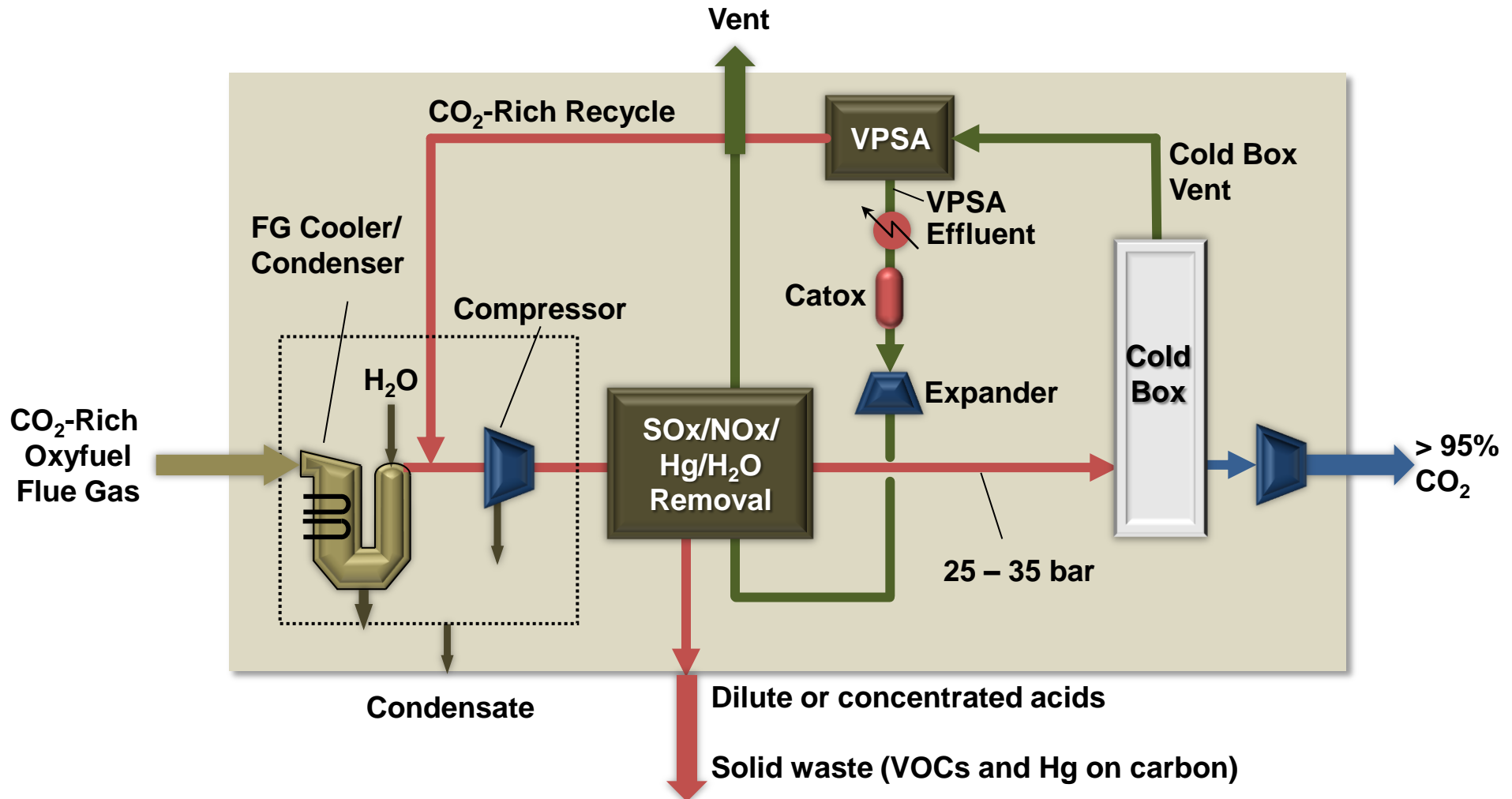
- ◆ **A Fortune 300 company with 2009 sales of \$9 Bn**
- ◆ **One of the largest industrial gases companies in the world**
- ◆ **Markets served**
 - Metals, Energy, Chemicals, Healthcare, Electronics, Manufacturing, Food and Beverage, Aerospace and various other markets
- ◆ **Major gas products**
 - Atmospheric gases: O₂, N₂, rare gases (Ar, Xe, Kr, Ne)
 - Process Gases: H₂, CO₂, He, acetylene
 - Specialty Gases
- ◆ **Experience with several technologies applicable to CCS**
 - Cryogenic, adsorption and polymeric-membrane air separation
 - Hydrogen production
 - Carbon dioxide purification and liquefaction
 - Oxy-fuel combustion

Project Overview

- ◆ **Goal: Develop a near-zero emissions oxy-combustion flue gas purification technology**
 - >95% CO₂ capture for existing plants with high air ingress
 - Produce high purity CO₂ by removing >90% of SO_x/NO_x/Hg
- ◆ **Total cost: \$5.4MM**
 - DOE \$3.24 MM
 - Praxair \$2.16 MM
- ◆ **DOE Project # NT0005341**
 - DOE Program manager – Mike Mosser
- ◆ **Project performance dates: 1/1/09 – 12/31/11**
- ◆ **Project participants**
 - Praxair
 - Foster Wheeler
 - AES
 - WorleyParsons Canada

Technology Fundamentals

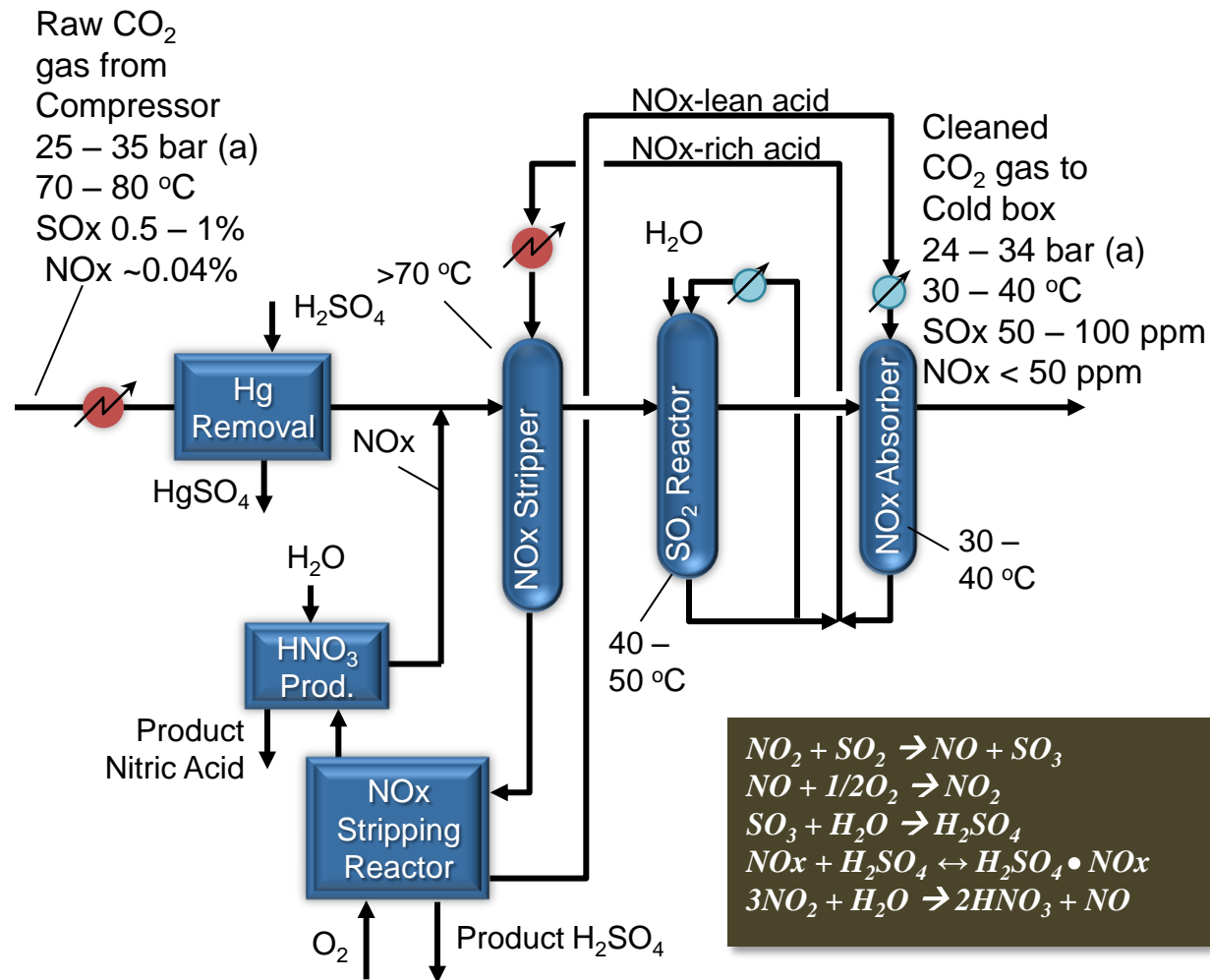
Near Zero Emissions CO₂ Processing Unit (CPU)



Technology Fundamentals

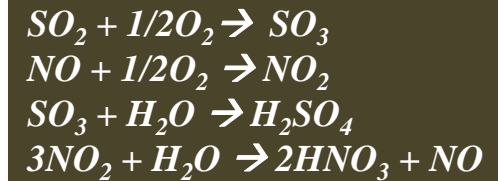
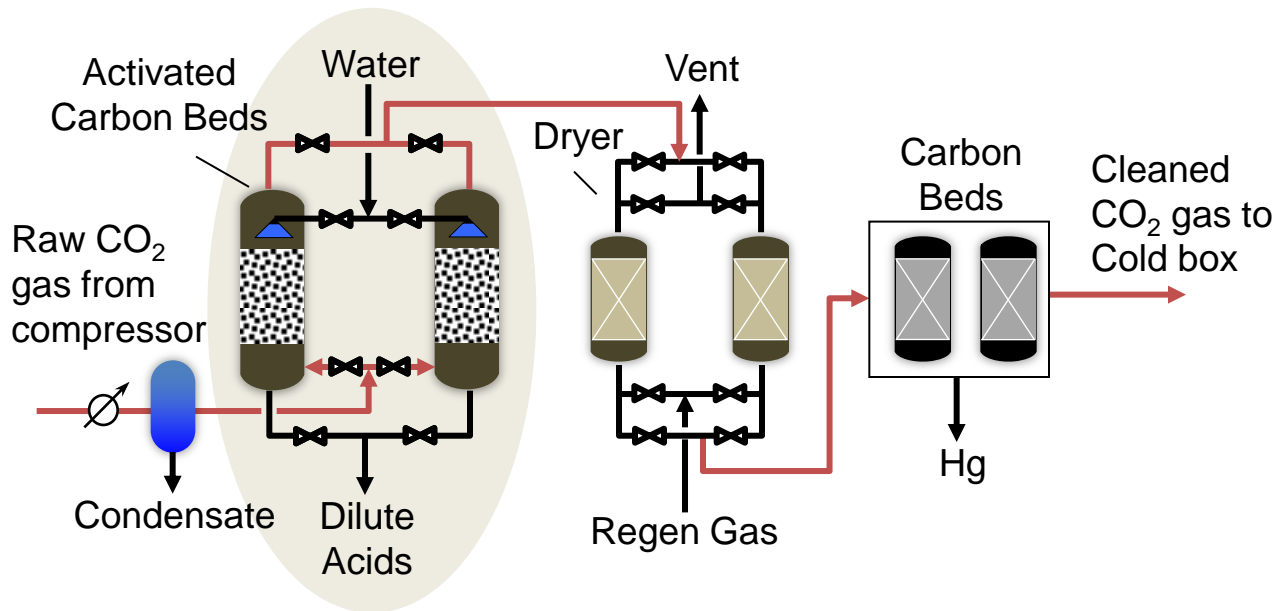
Sulfuric Acid Process for SO_x/NO_x/Hg Removal

- ◆ Modified lead chamber process
- ◆ SO_x and NO_x are converted to saleable acids
- ◆ Recirculation of NO_x using NO_x absorber and NO_x stripper aids the SO₂ oxidation reaction



Technology Fundamentals

Activated Carbon Process for SO_x/NO_x/Hg Removal

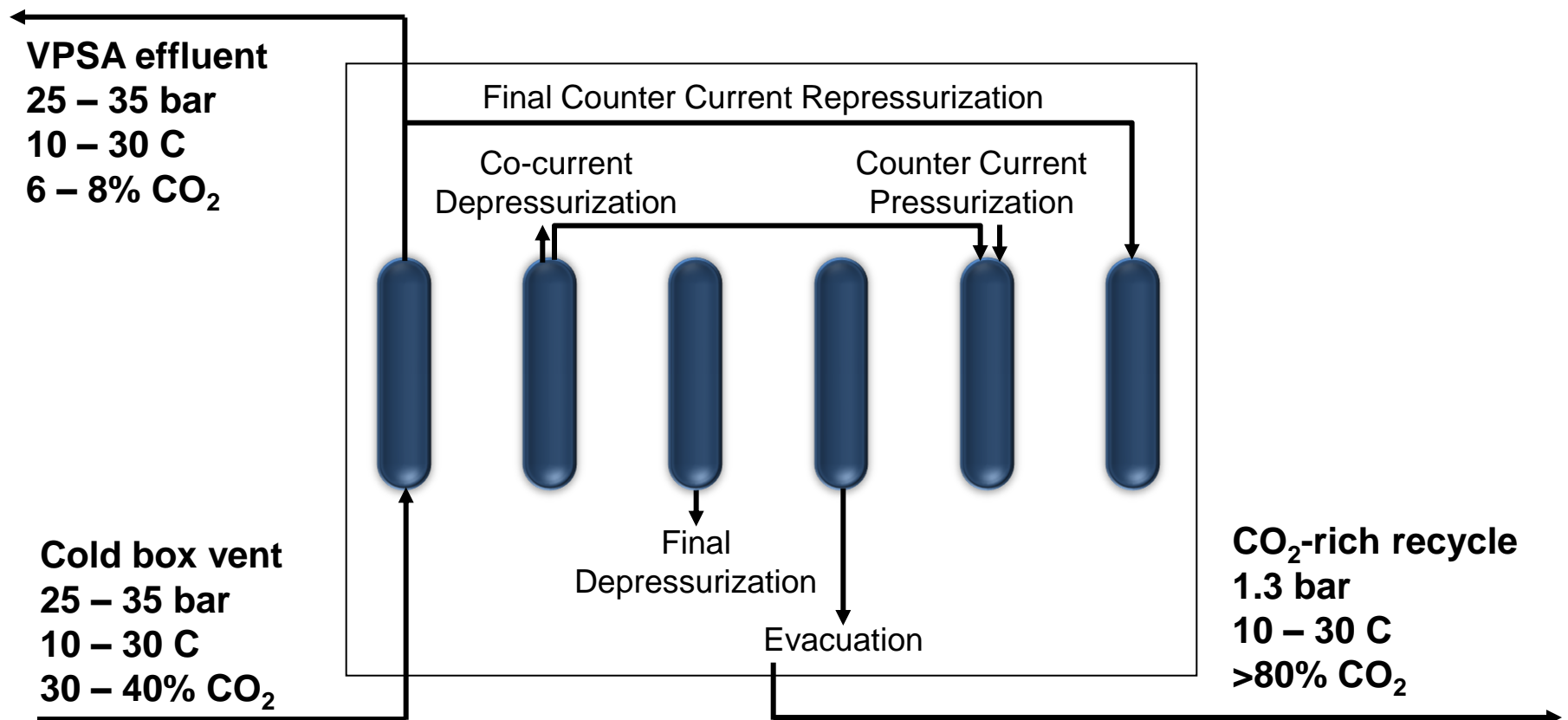


- ◆ SO₂ and NO are oxidized and retained on activated carbon
- ◆ Carbon is regenerated by water wash followed by drying
- ◆ Dilute acid stream is produced

Technology Fundamentals

VPESA (Vacuum Pressure Swing Adsorption)

- ◆ Multi-bed adsorption unit for separating CO₂ from cold box vent stream
- ◆ Simple cycle with minimum rotating equipment
- ◆ Shallow evacuation level



Benefits of Proposed Technologies

- ◆ **Near zero stack emissions**
- ◆ **High CO₂ recovery**
 - >95% for old plants with 10% air ingress
 - >99% for new plants with 2% air ingress
- ◆ **High CO₂ purity**
- ◆ **Sulfuric acid process**
 - Lower FGD/SCR operating costs for existing plants
 - Reduce/eliminate limestone, power, gypsum disposal and ammonia costs
 - Generate revenue from by-products
 - Lower capital and operating costs for new plants
 - Much smaller vessel sizes
- ◆ **Activated carbon process**
 - Lower investment costs for new plants
- ◆ **Lower CO₂ capture costs; Maximum benefit realized when**
 - Existing plant does not have FGD/SCR and high purity CO₂ is desired → alternative option will require installation of FGD/SCR
 - Existing plant has high air ingress → alternative option will have poor CO₂ recovery

Key Challenges

◆ Sulfuric acid process

- Maximum allowable SO_x in boiler may significantly limit flue gas SO_x levels
- High pressure & high temperature
- Technology development
- Material of construction
- Acid sales revenue will depend on product quality & proximity to customer

◆ Activated carbon process

- Longevity of process performance
- Disposal of dilute acid stream

◆ VPSA

- Tolerance to residual SO_x/NO_x in cold box vent

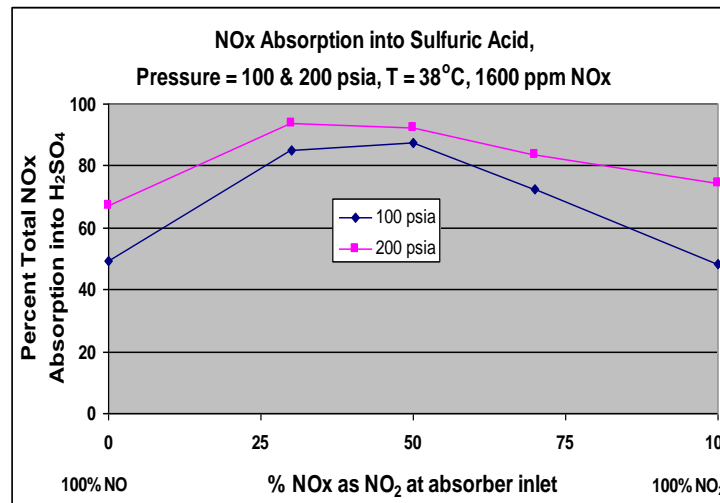
◆ Commercial viability

- Low efficiency of existing plants
- Capital cost advantage compared to conventional technology
- Adoption by power producers

Current Status

Sulfuric Acid Process

- ◆ **Bench-scale single column (1' L, 1.5" ID) unit**
 - Capacity – 0.06 tpd CO₂ in flue gas
 - Capable of testing different unit ops of the process
 - Synthetic flue gas contacted with circulating H₂SO₄
- ◆ **Gas phase nitric oxide (NO) oxidation kinetics confirmed**
- ◆ **NOx mass transfer in H₂SO₄ is being evaluated**
- ◆ **>90% NOx absorption in one stage; higher pressure and NO:NO₂ ratio close to 1:1 improved NOx absorption**
- ◆ **NOx removal from acid may be challenging**



Current Status

Activated Carbon Process

- ◆ **Two carbon materials selected based on SO_x removal screening tests**
- ◆ **Bench-scale unit with one carbon bed (1' L, 1" ID) built**
 - Capacity – 0.02 tpd CO₂ in flue gas
 - Synthetic flue gas is fed until breakthrough of SO_x or NO_x
- ◆ **Excellent simultaneous SO_x/NO_x removal achieved**
 - SO₂ >99 % and NO_x >96 %
- ◆ **Performance enhanced by**
 - Lower temperature
 - Higher pressure
 - Presence of moisture

Current Status

VP SA

- ◆ **Bench-scale unit (0.03 tpd) built for screening adsorbents**
 - Three adsorbents selected based on cost, CO₂ recovery, CO₂ purity and vacuum pump size
- ◆ **Pilot unit with 12 vessels (L ~ 11', ID ~ 2.5") commissioned**
 - Capacity – cold box vent containing 0.3 tpd CO₂ (equiv. to 3 tpd CO₂ in FG)
- ◆ **First set of data meets/exceeds the VP SA performance targets**
 - 99% capture rate with VP SA + cold box



Current Status

Commercial Viability

◆ Design basis

- 460 MW (gross) subcritical plant
 - Site ambients: 11.3 psia, 87 F, 26% RH
 - Bituminous (high sulfur) and PRB (low sulfur) coals
 - Existing FGD and SCR for SO_x/NO_x control
 - 2% air ingress
- Air separation unit producing 97% O₂
- Two CO₂ processing unit (CPU) designs
 - CO₂ purified to >95% purity and compressed to 153 bar
 - Acid Process for Bituminous-derived flue gas
 - Activated carbon process for PRB-derived flue gas
 - VPSA included in both CPUs

◆ Results

- > 99% reduction in stack emissions of CO₂ & pollutants and production of high purity CO₂ while reducing CO₂ capture costs by \$1 - \$3/ton compared to a conventional oxyfuel CO₂ purification process
- Power plant efficiency drops by ~10 percentage points in both the cases

Current Status

Performance Projections – Near Zero Emissions

Coal	Bituminous				PRB			
SOx/NOx removal	Sulfuric Acid Process				Activated Carbon Process			
	Composition by volume			% Reductions in stack emissions	Composition by volume			% Reductions in stack emissions
Component	Flue gas	Vent stream	Product CO ₂		Flue gas	Vent stream	Product CO ₂	
CO ₂	68.53 %	6.93%	96.93 %	98.8%	62.05 %	7.26%	96.92 %	98.9%
N ₂ + O ₂ + Ar	12.73%	92.86%	3.05%		10.58%	91.96%	3.08%	
H ₂ O	18.25 %	Nil	1 ppm		27.28 %	0.55%	1 ppm	
CO	284 ppm	<10 ppm	71 ppm	>99.5%	280 ppm	<10 ppm	83 ppm	>99.5%
SO _x	3884 ppm	Nil	68 ppm	>99.9%	471 ppm	Nil	7 ppm	>99.9%
NO _x	391 ppm	9 ppm	57 ppm	99.5%	156 ppm	7 ppm	5 ppm	99.6%
HCl	402 ppm	Nil	0 ppm	>99.9%	18 ppm	Nil	0 ppm	>99.9%
VOC	1.2 ppm	Nil	<0.1 ppm	>99.9%	1.3 ppm	Nil	<0.1 ppm	>99.9%
Hg	1.0 ppb	Nil	<0.1 ppb	>99.9%	10.1 ppb	Nil	<0.1 ppb	>99.9%

Stack Flow is ~98% (by wt.) Lower than Air-fired Operation

Future Plans

◆ Sulfuric Acid Process

- Complete bench scale tests (Q4 2010) and issue a topical report (Q1 2011)
- After successful completion, propose a new R&D program to conduct tests in a bench-scale unit (~0.1 tpd) that integrates all unit operations

◆ Activated Carbon Process

- Complete long-term regenerability tests (Q4 2010)
- Build and operate a dual bed continuous unit 2-5X the current unit (2011)

◆ VPSA

- Bench-scale tests for SO_x/NO_x tolerance (Q2 2011)
- Complete pilot tests (Q3 2011)
- Develop a simulation tool to predict process performance (Q3 2011)

◆ Commercial viability

- Technoeconomic analysis and operability assessment (Q3 2011)

◆ Proposed commercialization timeline

- 10 – 50 tpd CPU demo with activated carbon & VPSA (2012 – 2013)
- Ready for larger size units in 2015

Summary

- ◆ **Promising results with activated carbon & VPSA processes**
- ◆ **Acid process next steps will be determined by year end**
- ◆ **Technology can achieve high CO₂ recovery, high purity CO₂ and near zero stack emissions while lowering capture costs**
- ◆ **Targeting 10 – 50 tpd demonstration in 2012 – 2013**

Acknowledgement & Disclaimer

- ◆ *Acknowledgment:* “This material is based upon work supported by the Department of Energy under Award Number DE-NT0005341.”
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